

SODIUM HYPOCHLORITE (AQUEOUS SOLUTION)

CHEMICAL NAME	Sodium hypochlorite
CAS NUMBER	7681-52-9
TECHNICAL REQUIREMENTS	Appearance.....transparent liquid, slightly yellow
	Active chlorine, g/l.....min. 150 PN-EN 901/ LA/2118*
	NaOH+Na ₂ CO ₃ as NaOH**, g/l.....max. 20 BN-87/6016-53
	Iron**, g/l.....max. 0,05 LA/2071* / ICP

* The method is available upon request

** Guaranteed value

GENERAL DATA	Molar mass74,42 g/mole
	Solubility in waterunlimited
	Other solventsalkali
	Density (20 °C)1,2 g/ml
	Odourcharacteristic, suffocating
	Freezing pointbelow -20 °C
	Decomposition pointat 25 °C oxygen is liberated at 35 °C chlorine is liberated

APPLICATION

Sodium hypochlorite has a very strong biocidal effect. It's main application is the disinfection of swimming pools, disinfection of water intakes and its treatment for consumption. Due to its oxidizing properties, the product is used in the chemical industry as an oxidizer of organic products in the chemical synthesis process, production of intermediates used in the pharmaceutical industry. In everyday life we found it in the form of commercially available textiles bleach. Sodium hypochlorite in combination with NaOH forms a strongly disinfectant and corrosive compound antifungal, which is used in dentifrice dental flushing. At a concentration of 70% it is used as antiseptic liquid.